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**IITJEE 2009 Paper 1: Questions and Solutions - Vidyamandir ...**  
IIT-JEE2009-Code-0-Questions and Solutions-Paper-I and II 3 4. The Henry's law constant for the solubility of N2 gas in water at 298 K is 1.0 × 10 5 atm. The mole fraction of N2 in air is 0.8. The number of moles of N2 from air dissolved in 10 moles of water at 298 K and 5 atm pressure is

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FIITJEE Solutions to IIT-JEE: Paper 2008. Code-03Time: 3 Hours Question paper format and Marking scheme: (i) The question paper consists of 3 parts (Part I : Mathematics, Part II : Physics, Part III : Chemistry).Each part has 4 sections. (ii) Section I contains 6 multiple choice questions. Each question has 4 choices (A), (B), (C) and (D), out of which only one is correct.

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