

# Atom Board Answers

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### **Atom Board Answers**

Answer: a. All the mass of an atom is concentrated in the nucleus because atom contains three subatomic particles like electron, proton and neutron. Out of which, nucleus present at the centre of an atom contains two subatomic particles that's protons and neutrons and the mass of nucleus is the sum of mass of protons and neutrons located at the centre of an atom.

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The Atom Board Protons - blue marbles. Neutrons - red marbles. Electrons - small marbles. Make each of the following atoms. Make sure you make the right element and isotope. Make sure you have the correct number of electrons in the correct orbits (also called levels or shells).. As you make each atom be sure to observe how each element looks.

### **chap 18 no 3**

Maharashtra Board Class 8 Science Chapter 5: BYJU'S Important Questions & Answers ... Answer: a. Atom: An atom is defined as the structural and functional unit of matter. It is the smallest particle of an element which retains its chemical identity in all the physical and chemical changes.

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Answer: Question 14. Explain the rule with example according to which electrons are filled in various energy levels, Answer: The maximum number of electrons that can be present in any shell or orbit of an atom is given by the formula  $2n^2$ , where  $n$  is the serial number of the shell. Therefore: K shell,  $n = 1$ , no. of electrons =  $2 \times 1^2 = 2$

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